Ageing effect on microstructural and mechanical properties of mullite-ZrO₂-TiO₂ **composites**

M. FATIMA MELO

LNETI/D TM Materials Technology Department, Lumiar, 1699 Lisbon (Codex), Portugal

J. S. M OYA

Instituto de Ceramica y Vidrio, C.S.I.C., Arganda del Rey, Madrid, Spain

The microstructural and mechanical properties of mullite-zirconia composites with $TiO₂$ (0.25 and 1.0mol) additions have been studied, after ageing the samples over a wide temperature range (1000 to 1500 $^{\circ}$ C) for long periods of time (100 to 200h). In the sample with 0.25 mol $TiO₂$ addition, changes in mullite composition and in the solid state compatibility at temperatures below 1450 $^{\circ}$ C were detected. In the sample containing 1 mol TiO₂, decomposition of AI₂TiO₅ occurs at $T \le 1200$ °C. Both compositions exhibit no increment in zirconia average grain size during ageing and, concomitantly, there is no strength degradation until higher temperatures (>1400°C) are reached, which become more drastic when $Al₂TiO₅$ is present.

1. **Introduction**

Mullite-zirconia reaction-sintered composites with $TiO₂$ additions have proved to be $[1-3]$ low-cost materials with reasonable high strength (300 to 350 MPa) and toughness (\approx 5 MPa m^{1/2}) values.

It has been shown in the literature [4-6] that higher temperature mechanical performance of zirconiabased ceramics is very dependent on microstructure as well as on microchemistry changes during ageing treatments. In order to understand the higher temperature microstructural and chemical stability of mullite-zirconia containing $TiO₂$ composites and how these parameters affect the mechanical behaviour, a study was made of the microstructural changes in these multicomponent composites as well as their effect on the mechanical properties during hightemperature ($\geq 1000^{\circ}$ C) ageing treatments.

2. Experimental procedure

Taking into account the compatibility relationships in the $ZrO_2-SiO_2-Al_2O_3-TiO_2$ system [7], two different kinds of composition were formulated, according to the equations and the following molar proportions

with $X = 0.25$

$$
2ZrSiO4 + (3 + X)Al2O3 + XTiO2
$$

\n
$$
\rightarrow 2ZrO2(ss)
$$
 + Al₆Si₂O₁₃(ss) (1)

$$
2ZrSiO4 + (3 + X)Al2O3 + XTiO2
$$

\n
$$
\rightarrow 2ZrO2(ss)
$$
 + Al₆Si₂O₁₃(ss) + Al₂TiO₅ (2)
\nwith $X = 1$

These are the (a) and (b) in Fig. 5.

Appropriate quantities of zircon (Zircon Opazir-S Quiminsa S.A., Spain), Al_2O_3 (Al_2O_3 CT 3000 SG

Alcoa, USA) and titania (Titania Merck, West Germany) required to produce mullite-zirconia and mullite-zirconia-aluminium titanate composites as indicated by Reactions 1 and 2, were used. In both cases the powders were homogenized by attrition milling in a teflon-coated attritor using alumina balls in an isopropyl alcohol suspension. After drying and sieving, the raw compositions were isostatically pressed at 200 MPa.

The resulting cylindrical green bars were sintered at 1500 \degree C, 2h (X = 1 composition) and 1550 \degree C 2h $(X = 0.25)$ in air. Sintered bars were subsequently aged at 1000, 1200, 1400, 1450 and 1500~ for times ranging from 0 to 200 h.

Powdered samples ($d < 35 \mu m$), after sintering and ageing, were then analysed by X -ray diffraction (XRD) using CuK α radiation.

The final density of the sintered and aged material was determined by the Archimeds method using distilled water.

The sample microstructure was observed by reflected light optical microscopy (RLOM) and by scanning electron microscopy-energy dispersive spectrometry (SEM-EDS) after polishing and thermal etching. The zirconia average grain size was estimated in samples submitted to different heat treatments, by measuring lineal intercepts on scanning electron micrographs according to the Fullman and Richmon technique [8].

Bending strength (σ_F) was determined by the threepoint bend method on cylindrical bars with \approx 4 mm diameter and ≈ 80 mm length. The toughness (K_{tc}) value of different samples was determined by the indentation method [9].

3. Results and discussion

In Fig. I the phase evolution plotted against ageing

Figure 1 Phase evolution plotted against ageing time at different temperatures for (a) $X = 0.25$, (b) $X = 1$. (O) Mullite (110), (\Box) $ZrSiO_4$ (200), (Δ) m-ZrO₂ (111), (\star) Al₂TiO₅ (110), (\bullet) rutile (I 1o).

time at different temperatures for $X = 0.25$ and $X = 1$ compositions is shown. The RLOM micrographs corresponding to $X = 0.25$ and $X = 1$ samples are given in Fig. 2. Scanning electron micrographs corresponding to $X = 0.25$ as-sintered and after 100 h ageing at 1200 \degree C, and $X = 1$ indicating the presence of zircon grains, are shown in Fig, 3. Fig. 4 shows the σ_f -time plots at different temperatures for $X = 0.25$ and $X = 1$ compositions.

The density and K_{IC} values for the different samples are reported in Table I. For $X = 1$ composition (Fig. 1) the only change observed in the initial starting phases, during ageing at different temperatures occurs at 1200° C as a consequence of aluminium titanate eutectoid decomposition into rutile plus alumina [10].

Conversely, when the composition is located in the solid solution region $(X = 0.25)$, the appearance of zircon during the ageing treatment at $T \le 1400^{\circ}$ C without the simultaneous appearance of α -Al₂O₃ could be related to the shift of mullite composition from that as-sintered (fired at 1550° C) towards others with high alumina content. The A1/Si ratio in the samples asfired (1550 \degree C, 2h) and after 100 h ageing at 1200 \degree C changes from 2.90 to 3.41 which is in agreement with previous statement (Fig. 3). Okada *et al.* [11] reported a large chemical composition change in mullite formed at different temperatures. The mullite formed at ≈ 1000 °C showed higher alumina content

Figure 2 Optical microstructure of both aged samplles (a) $X = 0.25$ composition as-sintered; (b) $X = 0.25$ composition aged at 1500°C, 100 h; (c) $X = 1$ composition as-sintered; (d) $X = 1$ composition aged at 1400°C, 200 h.

Figure 3 Scanning electron micrographs and corresponding EDX spectra for $X = 0.25$ composition, (a) as-sintered, A/Si = 2.90; (b) aged at 1200° C, 100 h, $A/Si = 3.41$.

in the sample $X = 0.25$ (Figs 3 and 5) at temperatures below 1450° C the new compatibility tetrahedra for this composition could be A_3S_2 (ss)- $Z_{(ss)}$ -ZS.

On the microstructural evolution of both samples during ageing, the following points can be made.

1. In both compositions, the porosity appears to be constant over all ageing treatments. No noticeable variation in the final density was detected (Table I).

2. A significant increase in the average pore size was detected in compositions $X = 0.25$ and $X = 1$ after ageing at 1450 and 1400° C, respectively. This effect was more evident in composition $X = 1$ (Fig. 2), consequently, pore coalescence takes place at these temperatures.

3. At ageing temperatures $\geq 1400^{\circ}$ C, zirconia and aluminium titanate (in $X = 1$) grain growth takes place, as can be observed in Figs 2 and 6.

The strength degradation observed for compositions $X = 0.25$ and $X = 1$ at $T \ge 1450^{\circ}$ C and

(75 mol%) than that obtained at $T > 1400^{\circ}$ C (65 mol $%$). The authors explained these results as a problem of metastability.

The results obtained in the present investigation supported the possibility that the equilibrium solid solution region in the $SiO_2 - Al_2O_3$ system [12] below 1200° C shift toward the higher alumina region.

Fig. 5 represents the different sections of the compatibility tetrahedra over the plane $\text{ZS-Al}_2\text{O}_3-\text{TiO}_2$ at $\approx 1600^{\circ}$ C. Considering the phase evolution observed

	Sample	As-sintered	1200° C. 200 _h	1400° C 200 _h	1450° C 200 _h	1500° C 100 _h
Final density	$X = 0.25$	3.65	3.64	3.66	3.66	3.63
$(g \, cm^{-3})$		(± 0.01)	(± 0.01)	(± 0.00)	(± 0.01)	(± 0.02)
	$X=1$	3.58	3.54	3.62		
		(± 0.02)	(± 0.11)	(± 0.01)		
$K_{\rm IC}$	$X = 0.25$	4.4	4.5	4.5	4.7	4.8
$(MPa m^{1/2})$		(± 0.1)	(± 0.2)	(± 0.2)	(± 0.3)	(± 0.4)
	$X = 1$	4.0	4.1	4.4		
		(± 0.2)	(± 0.2)	(± 0.2)	\overline{a}	

TABLE I Final density and toughness values

Figure 4 Bending strength plotted against time at different ageing treatments for (a) $X = 0.25$, (b) $X = 1$. (O) As-sintered, (\bullet) 1000° C, (Δ) 1200° C, (\Box) 1400° C, (\star) 1450° C, (\blacktriangledown) 1500° C.

 $T \geq 1400^{\circ}$ C, respectively, can be associated with pore coalescence and zirconia particle coarsening. These effects are more marked in sample $X = 1$, where, in addition, aluminium titanate grain growth is also present to produce a negative effect in the $\sigma_{\rm f}$ value (Fig. 4). K_{IC} increases in the samples treated at temperatures $\geq 1400^{\circ}$ C for both compositions as shown in Table I. This is probably due to the contribu-

Figure 5 Compatibility tetrahedra sections corresponding to $\approx 1600^{\circ}$ C with the inherent solid solutions. 1, A₃S_{2ss}-Z_{ss}; 2, Z_{ss} -A₃S_{2ss}-A; 3, Z_{ss} -A₃S_{2ss}-A-AT_{ss}; 4, Z_{ss} -A₃S_{2ss}-AT_{ss}; 5, A₃S_{2ss}- Z_{ss} -ZS.

Figure 6 Plot of average grain size against temperature. (Δ) $X = 0.25$, (O) $X = 1$, after 100 h ageing.

tion of microcracking-toughening as a consequence of the higher average grain size of zirconia particles. This fact agrees with data previously reported by Srikrishna et al. [12].

4. Conclusions

1. No microstructural or mechanical degradation is observed for $X = 0.25$ and $X = 1$ compositions at temperatures below 1450 and 1400°C, respectively.

2. Degradation is present in both composites at temperatures higher than 1450 and 1400° C and is related to pore coalescence and particle coarsening. Both effects are more evident in composition $X = 1$.

3. Because of the change in mullite composition and in solid state compatibilities which take place at $T < 1450^{\circ}$ C, the zircon is compatible with mullite (ss) and zirconia (ss), in $X = 0.25$ composition. This change does not substantially affect the mechanical properties of the samples.

4. An increase in K_{IC} values is seen for both composites during ageing treatments at higher temperatures (> 1400 °C).

Acknowledgements

The authors thank P. Pena for helpful discussions and C. Victor, A.M. Geraldes and T. Magalhães for experimental assistance.

References

- M. F. MELO, J. S. MOYA, P. PENA and S. DE AZA, J. Mater. Sci. 20 (1985) 2711.
- M. F. MELO and J. S. MOYA, Bol. Soc. Esp. Ceram. Vidr. $26(3)$ (1987) 163
- J. RINCON, J. S. MOYA and M. F. MELO, Br. Ceram. 3. Trans. J. 85 (1986) 201.
- 4. TAKAKI MASAKI and YUKIO MURATA, J. Mater. Sci. 22 (1987) 407.
- 5. SAN-YUAN CHEN and HONG-YANG LU, *ibid.* 23 ([988) 1195. "
- 6. K. URABE, J. MONMA, H. IKAWA and S. UDAGAWA, *Sci. Ceram.* 14 (1988) 503.
- 7. P. PENA and S, DE *AZA, ibid.* 12 (1983) 201.
- 8. C. RICHMON, Thesis, University of Sheffield (1971).
- 9. P. MIRANZO and S. MOYA, *Ceram. Int.* 10(4) (1984) 147.
- 10. P. PENA, S. DE AZA and J. S. MOYA, *Sci. Ceram.* 14 (1988) 751.
- II. K. OKADA and N. OTSUKA, *ibid.* 14 (1988) 497.
- I2. K. SRIKRISHNA, G. THOMAS and J. S. MOYA, in "Zirconia 86 Third International Conference on Science and Ceramics-Metal Systems" (Plenum, New York, 1986) pp. 155-63.

Received 13 January and accepted 24 August 1989

 \sim